

CONTENTS

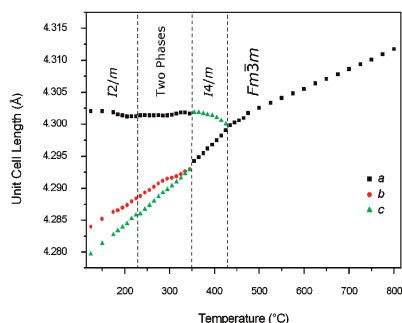
Abstracted/indexed in BioEngineering Abstracts, Chemical Abstracts, Coal Abstracts, Current Contents/Physics, Chemical, & Earth Sciences, Engineering Index, Research Alert, SCISEARCH, Science Abstracts, and Science Citation Index. Also covered in the abstract and citation database SCOPUS[®]. Full text available on ScienceDirect[®].

Regular Articles

Structural phase transitions and crystal chemistry of the series $Ba_2LnB'O_6$ (Ln = lanthanide and $B' = Nb^{5+}$ or Sb^{5+})

Paul J. Saines, Brendan J. Kennedy and Margaret M. Elcombe

Page 401

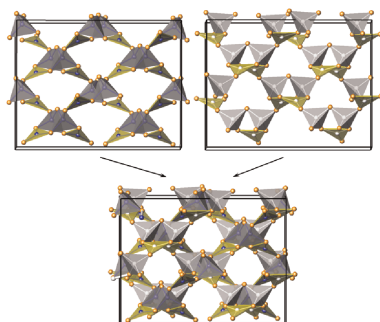


Lattice parameters versus temperature for Ba_2NdNbO_6 . The formation of the $I4/m$ tetragonal phase contrasts with the antimonate series where a rhombohedral structure occurs instead. This difference is believed to be caused by the presence of π -bonding present in the niobates but absent in the antimonates.

Crystal structure and phonon properties of noncentrosymmetric $LiNaB_4O_7$

M. Mączka, A. Waśkowska, A. Majchrowski, J. Żmija, J. Hanuza, G.A. Peterson and D.A. Keszler

Page 410



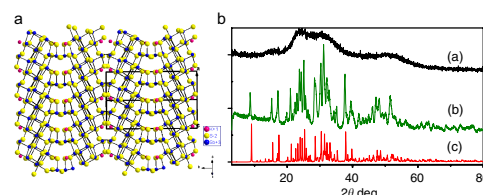
Structure of $LiNaB_4O_7$ borate showing two independent B_4O_7 frameworks and their interpenetration.

Regular Articles—Continued

Local structure and influence of bonding on the phase-change behavior of the chalcogenide compounds $K_{1-x}Rb_xSb_5S_8$

J.B. Wachter, K. Chrissafis, V. Petkov, C.D. Malliakas, D. Bilc, Th. Kyratsi, K.M. Paraskevopoulos, S.D. Mahanti, T. Torbrügge, H. Eckert and M.G. Kanatzidis

Page 420

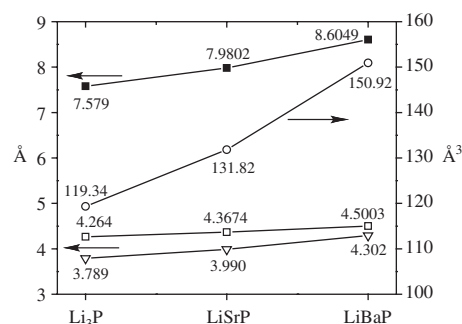


The KSb_5S_8 is a good example of a phase-change material with a mixed ionic/covalent bonding. The members of the $K_{1-x}Rb_xSb_5S_8$ series exhibit phase-change properties with greater glass forming ability (GFA) than KSb_5S_8 . The GFA increases with increasing Rb content. In this case, the random alloy disorder in the alkali metal sublattice seems to predominate over the increased degree of ionicity in going from $K \cdots S$ to $Rb \cdots S$ bonding and works to stabilize the glass forms in $K_{1-x}Rb_xSb_5S_8$.

Synthesis and single crystal structures of ternary phosphides Li_4SrP_2 and $AAeP$ ($A = Li, Na$; $Ae = Sr, Ba$)

Yongkwan Dong and Francis J. DiSalvo

Page 432

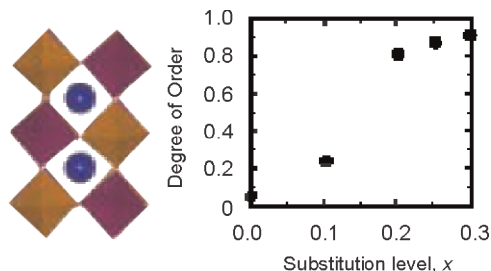


Comparison of the unit cell parameters and anionic interlayer distances between Li_3P and $LiAeP$ ($Ae = Sr, Ba$). (\square): a -axis, (\blacksquare): c -axis, (\circ): unit cell volume, (∇): interlayer distance).

Metal valences in electron-doped (Sr,La)₂FeTaO₆ double perovskite: A ⁵⁷Fe Mössbauer spectroscopy study

E.-L. Rautama, J. Lindén, H. Yamauchi and M. Karppinen

Page 440

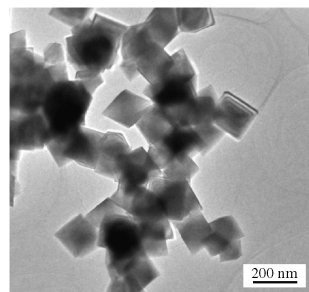


Both the degree of order and the valence states of Fe and Ta are controlled in the (Sr_{1-x}La_x)₂FeTaO₆ double perovskite oxide through aliovalent La^{III}-for-Sr^{II} substitution.

High-yield synthesis and characterization of monodisperse sub-microsized CoFe₂O₄ octahedra

Xian-Ming Liu, Shao-Yun Fu and Lu-Ping Zhu

Page 461



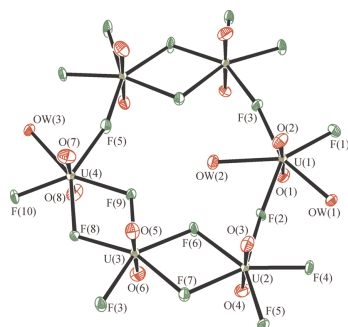
TEM image of CoFe₂O₄ octahedra

Sub-micro-sized CoFe₂O₄ octahedra with a high yield are synthesized via a simple hydrothermal route under mild conditions. TEM image shows that the edge size of CoFe₂O₄ octahedra is estimated to be about 0.10–0.14 μm.

Hydrothermal synthesis, crystal structure, and characterization of a new pseudo-two-dimensional uranyl oxyfluoride, [N(C₂H₅)₄]₂[(UO₂)₄(OH₂)₃F₁₀]

Kang Min Ok and Dermot O'Hare

Page 446

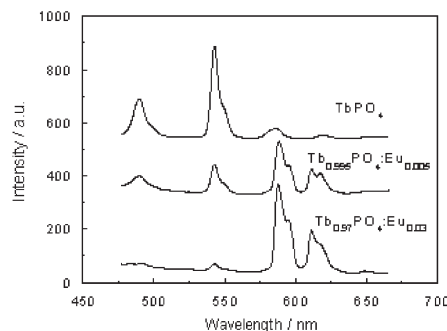


ORTEP (50% probability ellipsoids) drawing of [N(C₂H₅)₄]₂[(UO₂)₄(OH₂)₃F₁₀] in the *ab*-plane. Note the pentagonal bipyramidal environments of U⁶⁺ cations to form a novel six-membered ring.

Energy transfer and heat-treatment effect of photoluminescence in Eu³⁺-doped TbPO₄ nanowires

Weihua Di, Xiaojun Wang, Peifeng Zhu and Baojiu Chen

Page 467



The energy transfer from Tb³⁺ to Eu³⁺ occurs in the Eu-doped TbPO₄ nanowires synthesized by the hydrothermal route.

Habit modification of calcium carbonate in the presence of malic acid

Zhaofeng Mao and Jianhua Huang

Page 453

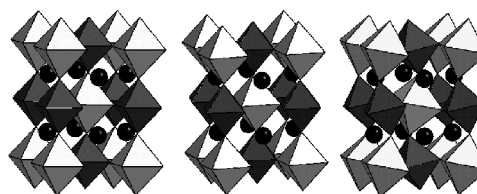


Dumbbell-like CaCO₃ particles obtained in the presence of malic acid.

Structural and magnetic study of the cation-ordered perovskites Ba_{2-x}Sr_xErMoO₆

Edmund J. Cussen

Page 474

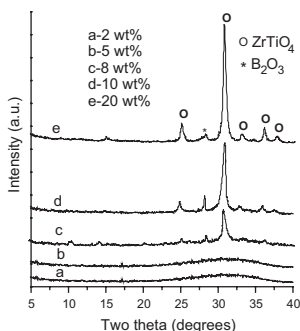


Introduction of Sr²⁺ into the double perovskite Ba₂ErMoO₆ introduces increasingly large distortions from cubic symmetry and permits antiferromagnetic order at T_N ≤ 4 K.

The effect of B₂O₃ addition on the crystallization of amorphous TiO₂-ZrO₂ mixed oxide

Dongsen Mao and Guanzhong Lu

Page 484

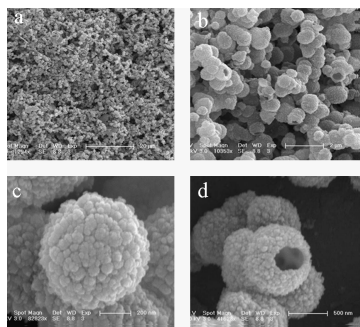


The addition of small amount of boria (<8 wt%) hinders the crystallization of amorphous TiO₂-ZrO₂ into a crystalline ZrTiO₄ compound, while the larger amount of boria (≥8 wt%) promotes the crystallization process.

Template free fabrication of hollow hematite spheres via a one-pot polyoxometalate-assisted hydrolysis process

Baodong Mao, Zhenhui Kang, Enbo Wang, Chungui Tian, Zhiming Zhang, Chunlei Wang, Yanli Song and Meiye Li

Page 489

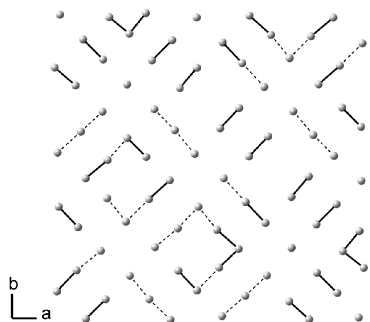


Uniform hollow hematite (α -Fe₂O₃) spheres with diameter of about 600–700 nm and shell thickness lower than 100 nm are obtained by direct hydrothermal treatment of dilute FeCl₃ and tungstophosphoric acid H₃PW₁₂O₄₀ solution at 180 °C. The hollow spheres present a high surface area and weak ferromagnetic behavior at room temperature.

The Pr₂Se₃-PrSe₂ system: Studies of the phase relationships and the modulated crystal structure of PrSe_{1.85}

Thomas Doert, Christian Graf, Peer Schmidt, Inga G. Vasilieva, Paul Simon and Wilder Carrillo-Cabrera

Page 496

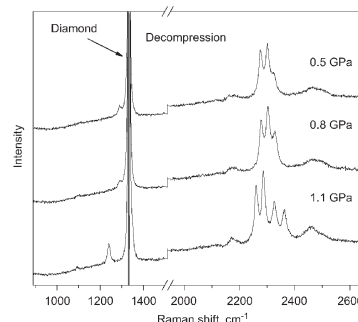


Section of the modulated Se layer of PrSe_{1.85}.

High-pressure phase transition in LiBH₄

A.V. Talyzin, O. Andersson, B. Sundqvist, A. Kurnosov and L. Dubrovinsky

Page 510

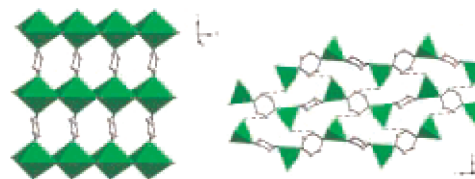


Raman spectra of LiBH₄ in the region of the phase transition recorded during decompression.

Structure characterization of 1:1 adducts of metal(II) halides and piperazine

Jie-Hui Yu, Qin Hou, Tie-Gang Wang, Xiao Zhang and Ji-Qing Xu

Page 518

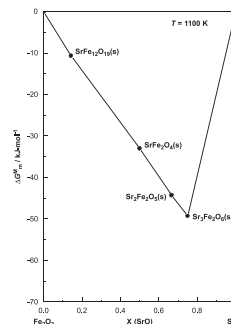


Two 1:1 adducts of MX₂ and piperazine (pip), [CdI₂(pip)] **1** and [CoCl₂(pip)] **2**, were prepared by the simple hydro/solvothermal techniques. With pip as the bridges, the 1-D linear CdI₂ chains are extended into a 2-D layered compound **1**, while the mononuclear CoCl₂ units are linked into a 1-D zigzag-type chain compound **2**. Compound **1** possesses the fluorescence property with the emission peaks at 373 nm and 410 nm.

Thermodynamic studies on SrFe₁₂O₁₉(s), SrFe₂O₄(s), Sr₂Fe₂O₅(s) and Sr₃Fe₂O₆(s)

S.K. Rakshit, S.C. Parida, S. Dash, Z. Singh, B.K. Sen and V. Venugopal

Page 523



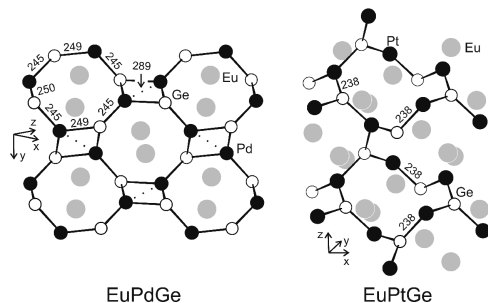
Gibbs energy of mixing of ternary oxides of the system Sr-Fe-O.

Continued

Chemical bonding in EuTGe ($T = \text{Ni, Pd, Pt}$) and physical properties of EuPdGe

Xavier Rocquefelte, Régis Gautier, Jean-François Halet, Ralf Müllmann, Carsten Rosenhahn, Bernd D. Mosel, Gunter Kotzbya and Rainer Pöttgen

Page 533

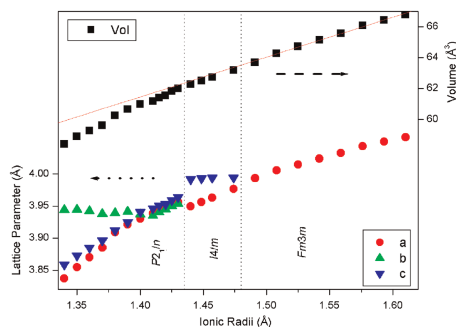


Cutouts of the [PdGe] and [PtGe] polyanions in the structures of EuPdGe and EuPtGe . Atom designations and some relevant interatomic distances are given.

Composition and temperature dependent phase transitions in Co-W double perovskites, a synchrotron X-ray and neutron powder diffraction study

Qingdi Zhou, Brendan J. Kennedy and Margaret M. Elcombe

Page 541

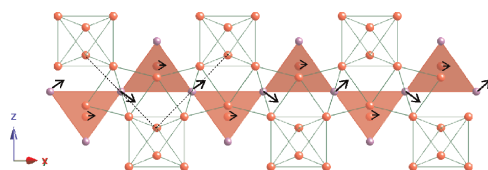


Variation of the reduced lattice parameters and volumes with the average A cation radius for the series $A_{2-x}\text{Sr}_x\text{CoWO}_6$. The reduced lattice parameters are, for $P2_1/n$ $a' = \sqrt{2}a_p$, $b' = \sqrt{2}b_p$, $c' = 2c_p$ for $I4/m$, $a' = \sqrt{2}a_p$, $c' = 2c_p$ and for $Fm\bar{3}m$ $d = 2a_p$.

The local crystal chemistry and dielectric properties of the cubic pyrochlore phase in the $\text{Bi}_2\text{O}_3\text{-M}^{2+}\text{O-Nb}_2\text{O}_5$ ($\text{M}^{2+} = \text{Ni}^{2+}$ and Mg^{2+}) systems

Binh Nguyen, Yun Liu and Ray L. Withers

Page 549

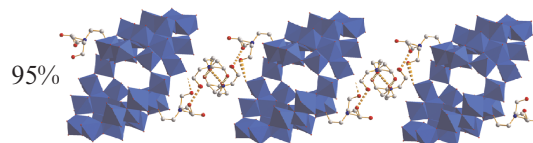


The characteristic β -cristobalite-type displacive disorder of the $\text{O}A_2$ sub-structure (of the ideal pyrochlore structure type) drawn relative to the surrounding $B_2\text{O}_6$ octahedral framework sub-structure and responsible for the observed structured diffuse intensity distribution in the $\text{Bi}_{1.65}\text{Ni}_{0.75}\text{Nb}_{1.50}\text{O}_7$ and $\text{Bi}_{1.67}\text{Mg}_{0.64}\text{Nb}_{1.53}\text{O}_7$ misplaced-displacive cubic pyrochlores.

A novel nanomolecular organic-inorganic hybrid compound: $\text{Na}_2[\text{NH}(\text{CH}_2\text{CH}_2\text{OH})_3]_4\{\text{Mo}_{36}\text{O}_{112}(\text{OH})_{14}[\text{OHCH}_2\text{CH}_2\text{NH}(\text{CH}_2\text{CH}_2\text{OH})_2]_2\} \cdot n\text{H}_2\text{O}$ ($n \approx 72$) exhibiting a supramolecular one-dimensional chainlike structure

Da-Dong Liang, Shu-Xia Liu, Chun-Ling Wang and Yuan-Hang Ren

Page 558

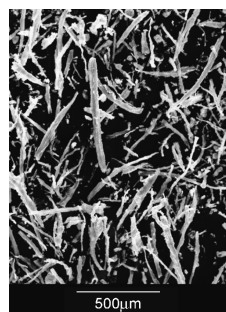


A novel organic-inorganic hybrid material based on macroisopolyanion $\{\text{Mo}_{36}\}$ has been synthesized and characterized by X-ray single-crystal crystallography, XRD, IR spectroscopy, and TG analysis. The material exhibits a supramolecular one-dimensional chainlike structure. The $\{\text{Mo}_{36}(\text{TEAH}^+)_2\}$ unit could be considered as nanosized chelating ligand with [2N, 4O] donor sets.

Mesoporous aluminosilicate ropes with improved stability from protozeolitic nanoclusters

Junlin Zheng, Dejin Kong, Weimin Yang, Zaiku Xie, Dong Wu and Yuhan Sun

Page 564

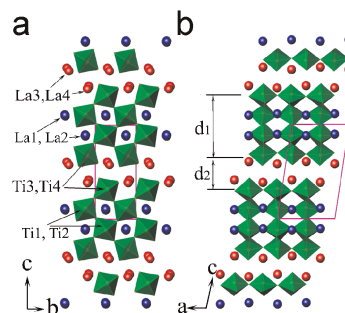


Mesoporous aluminosilicate ropes with enhanced hydrothermal stability were fabricated from protozeolitic nanoclusters through $\text{S}^+\text{X}^-\text{I}^+$ route in HNO_3 solution under the direction of CTAB templates.

Structural change of layered perovskite $\text{La}_2\text{Ti}_2\text{O}_7$ at high pressures

F.X. Zhang, J. Lian, U. Becker, R.C. Ewing, L.M. Wang, Jingzhu Hu and S.K. Saxena

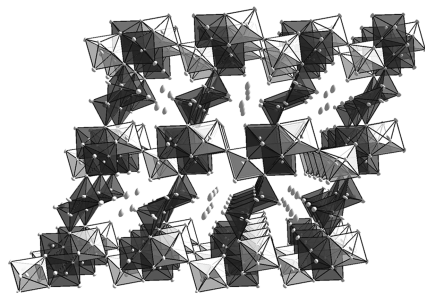
Page 571



$\text{La}_2\text{Ti}_2\text{O}_7$ is a layered perovskite and it shows anisotropic compressibility at high pressures. The in situ XRD and Raman measurements indicate a structural transformation takes place at 16.7 GPa.

Low-alkali metal content in β -vanadium mixed bronzes: The crystal structures of β - $K_x(V,Mo)_6O_{15}$ ($x = 0.23$ and 0.32) by single-crystal X-ray diffraction

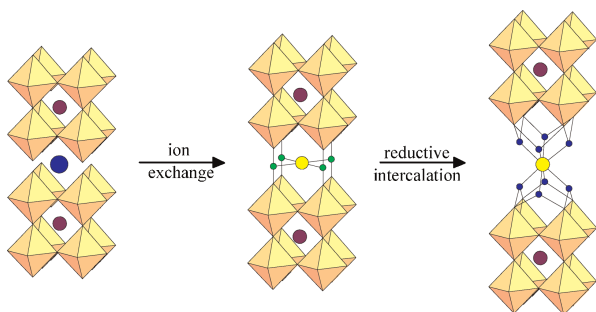
Michele Zema, Paolo Ghigna and Serena C. Tarantino
Page 577



Perspective view of the crystal structure of $K_xV_{6-y}Mo_yO_{15}$ along [010]. This vanadium-molybdenum mixed bronze assumes the β - $Na_xV_6O_{15}$ (bannermanite-type) structure, in which Mo(VI) substitutes for V only in the V2 position. The alkali metal content in the studied crystals is much lower than the solubility limit found for the analogous Na containing compounds.

Construction of a double-layered tetrahedral network within a perovskite host: Two-step route to the alkali-metal-halide layered perovskite, $(Li_xCl)LaNb_2O_7$

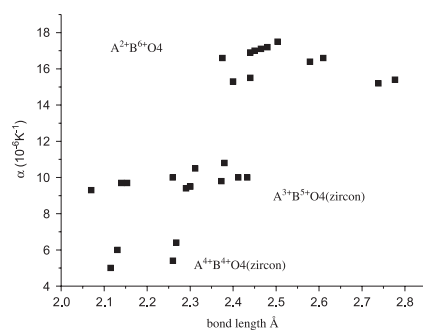
Liliana Viciu, Thomas A. Kodenkandath and John B. Wiley
Page 583



Alkali-halide layers can be constructed within a layered perovskite host via a two-step topochemical process.

The relationship between the thermal expansions and structures of ABO_4 oxides

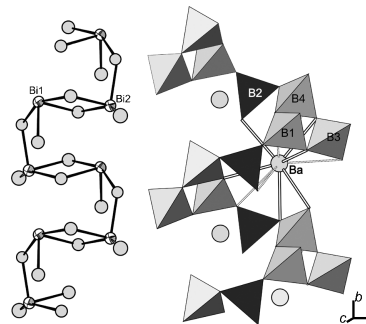
Huiling Li, Shihong Zhou and Siyuan Zhang
Page 589



The evaluated CTEs of A-O bonds change against A-O bond distances.

Preparation, crystal structure and thermal expansion of a new bismuth barium borate, $BaBi_2B_4O_{10}$

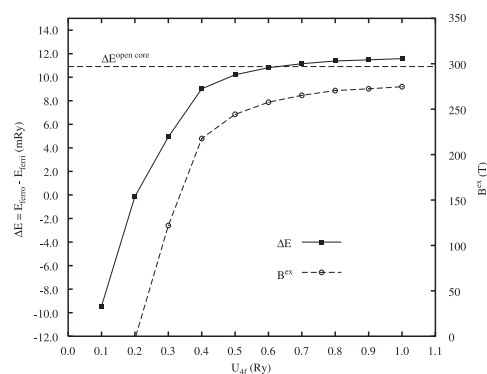
R.S. Bubnova, S.V. Krivovichev, S.K. Filatov, A.V. Egorysheva and Y.F. Kargin
Page 596



New borate and bismuthate one-dimensional polyions.

Magnetic properties of $DyCo_5$ and $TbCo_5$ intermetallics from the electronic structure calculations

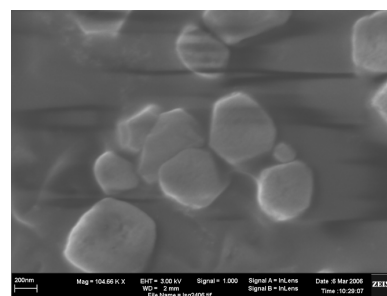
G.I. Miletic and Ž. Blažina
Page 604



Differences between total energies of ferromagnetic and ferrimagnetic configurations in $DyCo_5$ together with the corresponding exchange fields vs. the U_{df} parameter.

A sol-gel route for the development of rare-earth aluminum borate nanopowders and transparent thin films

Lauro J.Q. Maia, Valmor R. Mastelaro, Sebastien Pairis, Antonio C. Hernandez and Alain Ibanez
Page 611

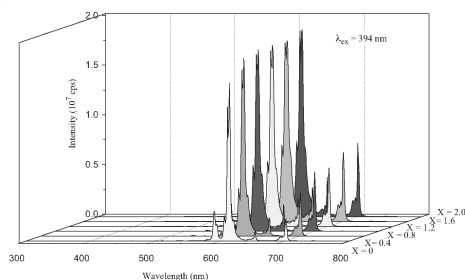


SEM micrograph of $Y_{0.9}Er_{0.1}Al_3(BO_3)_4$ powders calcined at 400 and 700 °C during 24 h and heat-treated at 1150 °C.

Continued

Structural, spectroscopic and photoluminescence studies of $\text{LiEu}(\text{WO}_4)_{2-x}(\text{MoO}_4)_x$ as a near-UV convertible phosphor
Chuang-Hung Chiu, Ming-Fang Wang, Chi-Shen Lee and Teng-Ming Chen

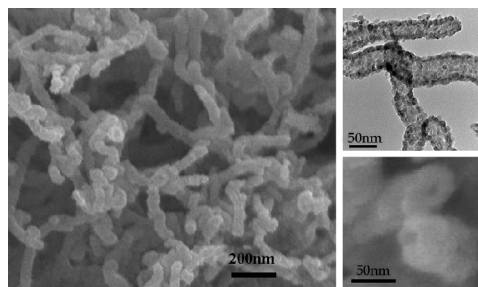
Page 619



As shown by the PL spectra of $\text{LiEu}(\text{WO}_4)_{2-x}(\text{MoO}_4)_x$ ($x=0, 0.4, 0.8, 1.2, 1.6, 2.0$) under 394 nm near-UV excitation, the intensity of $^5D_0 \rightarrow ^7F_2$ transition was found to increase with the increasing x and the $I-\lambda$ curve reaches a maximum when the relative ratio of Mo/W is 2:0.

Carbon nanotube assisted synthesis of CeO_2 nanotubes
Dengsong Zhang, Hongxia Fu, Liyi Shi, Jianhui Fang and Qiang Li

Page 654

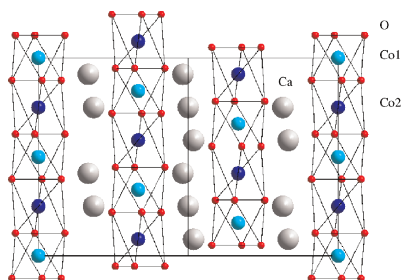


CeO_2 nanotubes have been synthesized facily using carbon nanotubes as templates by a liquid phase deposition method. The obtained CeO_2 nanotubes are composed of many tiny interconnected nanocrystallites of about 10 nm in size, and have a uniform diameter ranging from 40 to 50 nm.

Structure and magnetism of rare-earth-substituted $\text{Ca}_3\text{Co}_2\text{O}_6$

Charles H. Hervoches, Helmer Fjellvåg, Arne Kjekshus, Vivian Miksch Fredenborg and Bjørn C. Hauback

Page 628

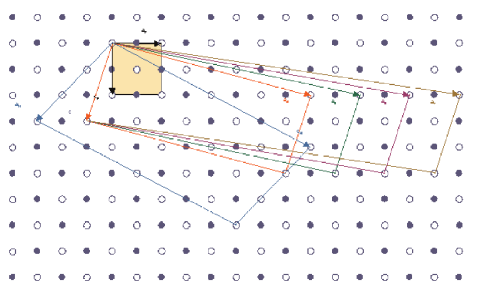


Yttrium- and rare-earth-substituted derivatives of $\text{Ca}_{3-v}\text{R}_v\text{Co}_2\text{O}_6$ have been synthesized. A significant increase in the Co2–O distances within the trigonal-prismatic Co_2O_6 coordination polyhedra confirms that extra electrons from the R^{3+} -for- Ca^{2+} substitution enter the Co2 site, thereby formally reducing its valence state. The long-range ferrimagnetic ordering of the parent $\text{Ca}_3\text{Co}_2\text{O}_6$ phase vanishes at $v > \sim 0.30$ in $\text{Ca}_{3-v}\text{R}_v\text{Co}_2\text{O}_6$.

$\text{Bi}_{2n+4}\text{Mo}_n\text{O}_{6(n+1)}$ with $n=3, 4, 5, 6$: A new series of low-temperature stable phases in the $m\text{Bi}_2\text{O}_3 - \text{MoO}_3$ system ($1.0 < m < 1.7$): Structural relationships and conductor properties

Eladio Vila, Ángel R. Landa-Canovas, Jean Galy, Juan E. Iglesias and Alicia Castro

Page 661

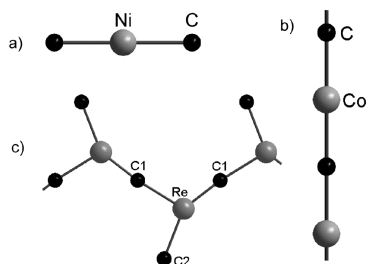


Four new phases $\text{Bi}_{10}\text{Mo}_3\text{O}_{24}$, $\text{Bi}_6\text{Mo}_2\text{O}_{15}$, $\text{Bi}_{14}\text{Mo}_5\text{O}_{36}$ and $\text{Bi}_8\text{Mo}_3\text{O}_{21}$ have been synthesized, characterized and described as consecutive members of a new family $\text{Bi}_{2n+4}\text{Mo}_n\text{O}_{6(n+1)}$, with $n=3, 4, 5$ and 6. They are based on basic fluorite-type structure and their unit cells are depicted in the accompanying graphic.

Ternary rare earth and actinoid transition metal carbides viewed as carbometalates

Enkhtsetseg Dashjay, Guido Kreiner, Walter Schnelle, Frank R. Wagner, Rüdiger Kniep and Wolfgang Jeitschko

Page 636

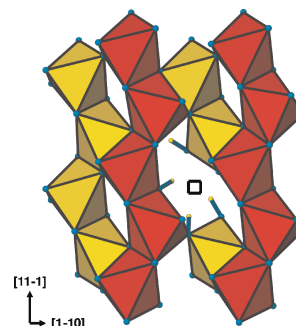


Carbometalates represent a specific class of ternary carbides whose crystal structures contain complex anions, e.g., (a) discrete complex anion $[\text{NiC}_2]^{8-}$ in the crystal structure of Th_2NiC_2 , (b) $[\text{CoC}]^{1-}$ linear chain in the crystal structure of YCoC , (c) $[\text{ReC}_2]^{6-}$ zigzag chain in Pr_2ReC_2 .

Titanium vacancy defects in sol-gel prepared anatase

Ian.E. Grey and Nicholas C. Wilson

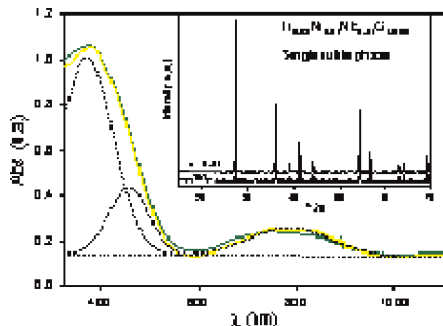
Page 670



Relaxed structure from DFT modelling of anatase containing one titanium vacancy and four protons.

Synthesis and characterization of $Ti_{1-2x}Nb_xNi_xO_{2-x/2}$ solid solutions

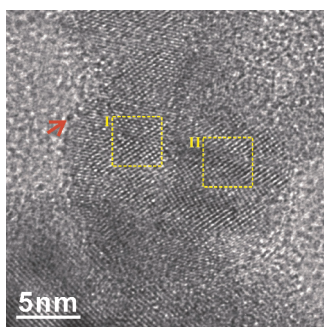
Mónica Martos, Beatriz Julián, Hakim Dehouli, Didier Gourier, Eloisa Cordoncillo and Purificación Escribano
Page 679



The synthesis and characterization of the $Ti_{1-2x}Nb_xNi_xO_{2-x/2}$ system prepared by traditional solid-state and sol-gel methodologies is reported. The incorporation of the doping ions in the rutile structure by Rietveld refinements and the magnetic response are discussed. The similarity found by both procedures introduces new possibilities of coloured TiO_2 -based solid solutions.

On the relaxation/transformation of NiO-dissolved TiO_2 condensates with fluorite-type derived structures

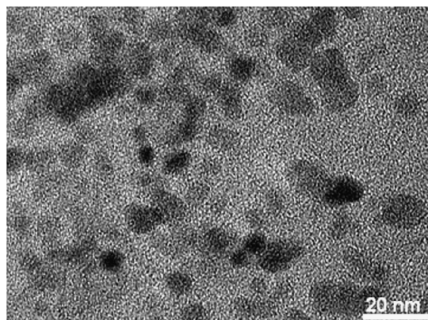
Chang-Ning Huang, Pouyan Shen and Shuei-Yuan Chen
Page 688



Lattice image of NiO-dissolved TiO_2 condensate having local fluorite-type (region I) and baddeleyite-type (region II) structures and (010) facet (arrow).

Hydrophilic and hydrophobic nano-sized Mn_3O_4 particles

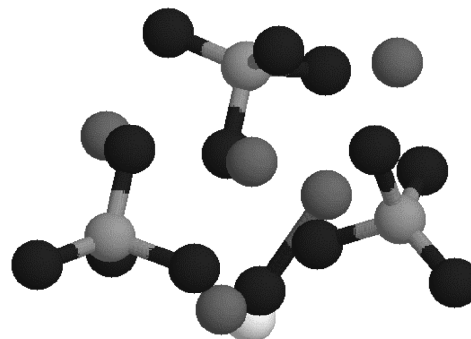
Pierre Gibot and Lydia Laffont
Page 695



Hydrophobic nano-sized Mn_3O_4 particles.

Quantum chemical and spectroscopic analysis of calcium hydroxyapatite and related materials

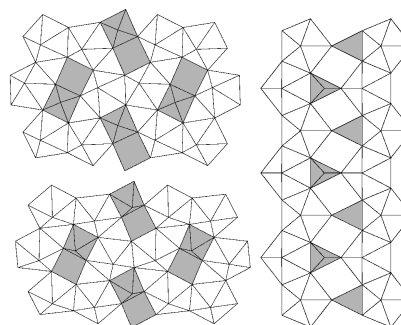
V.D. Khavryuchenko, O.V. Khavryuchenko and V.V. Lisnyak
Page 702



The structure of the quantum chemically optimized $Ca_5(PO_4)_3(OH)$ cluster, which was used for vibrations spectra simulation.

Hydrothermal synthesis, structure and thermal stability of diamine templated layered uranyl-vanadates

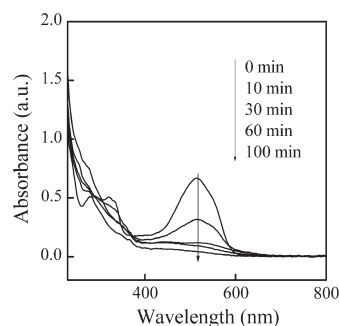
Murielle Rivenet, Nicolas Vigier, Pascal Roussel and Francis Abraham
Page 713



The three types of layer in layered uranyl-vanadates using diamine as a structure-directing agent.

Preparation and visible-light photocatalytic activity of Ag_3VO_4 powders

Xuexiang Hu and Chun Hu
Page 725

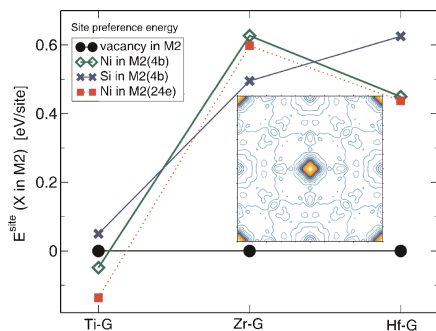


The UV/vis spectral changes of ARB solution recorded for NiO/Ag_3VO_4 as a function of irradiation time.

Continued

Crystal chemistry of the G-phases in the {Ti, Zr, Hf}-Ni-Si systems

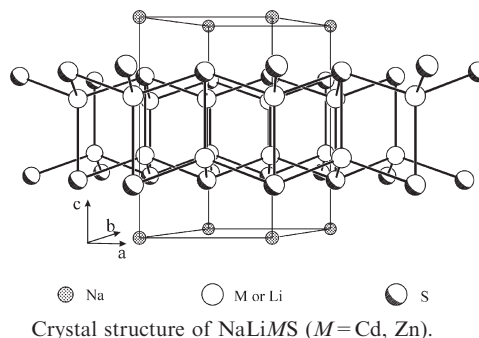
A. Grytsiv, Xing-Qiu Chen, P. Rogl, R. Podloucky, H. Schmidt, G. Giester and V. Pomjakushin
 Page 733



$\text{Ti}_6\text{Ni}_{16.7}\text{Si}_7$ (G-Phase, Ti-G) was found to crystallize in a new variant of the filled $\text{Th}_6\text{Mn}_{23}$ -type structure with an additional Ni atom partially occupying the 24e site inside Ti octahedra, where Ti atoms occupy a split position.

Synthesis, structure, optical properties, and electronic structure of NaLiCdS_2

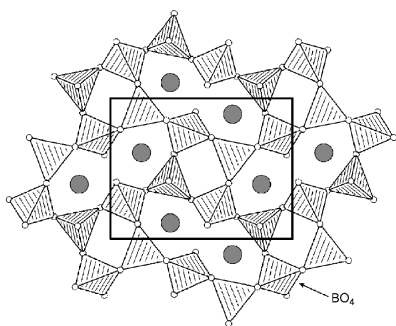
Bin Deng, George H. Chan, Fu Qiang Huang, Danielle L. Gray, Donald E. Ellis, Richard P. Van Duyne and James A. Ibers
 Page 759



Crystal structure of NaLiMS ($M = \text{Cd}, \text{Zn}$).

High-pressure synthesis, crystal structure, and properties of the first ternary hafniumborate $\beta\text{-HfB}_2\text{O}_5$

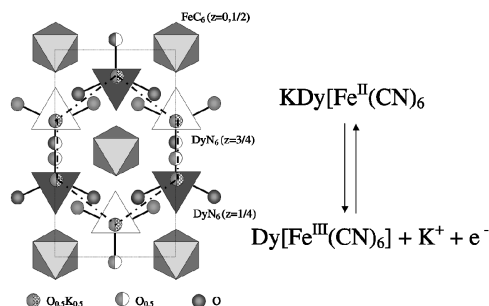
Johanna S. Knyrim and Hubert Huppertz
 Page 742



Synthesis of $\beta\text{-HfB}_2\text{O}_5$ via the multi-anvil high-pressure/high-temperature technique, representing the first ternary compound in the borate system Hf-B-O.

Charge transport properties in microcrystalline $\text{KDyFe}(\text{CN})_6$

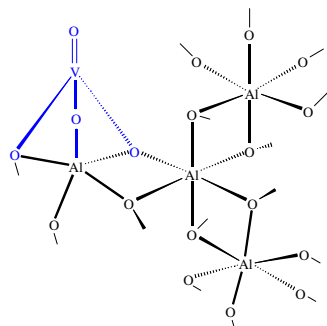
P.H. Aubert, F. Goubard, C. Chevrot and A. Tabuteau
 Page 765



Dy and Fe polyhedra packing in the cell of $\text{KDyFe}(\text{CN})_6 \cdot 3.5\text{H}_2\text{O}$ shows occluded water molecules and potassium ions forming a pseudo-hexagonal 2D sub-lattice connected to each other by diffusion channels.

Structural study of VO_x doped aluminium fluoride and aluminium oxide catalysts

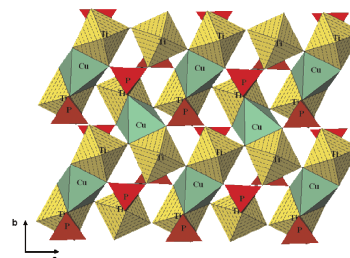
Kerstin Scheurell, Gudrun Scholz and Erhard Kemnitz
 Page 749



Structure model for VO_x doped aluminium oxide.

Synthesis, structure, and physicochemical investigations of the new $\alpha\text{-Cu}_{0.50}\text{TiO}(\text{PO}_4)$ oxyphosphate

S. Benmokhtar, H. Belmal, A. El Jazouli, J.P. Chaminade, P. Gravereau, S. Pechev, J.C. Grenier, G. Villeneuve and D. de Waal
 Page 772

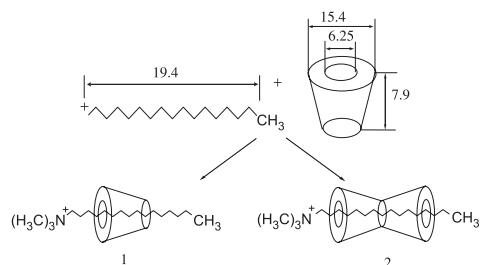


Projection of the structure of $\alpha\text{-Cu}_{0.50}\text{TiO}(\text{PO}_4)$ along the a -axis.

Fabrication of MCM-41 mesoporous silica through the self-assembly supermolecule of β -CD and CTAB

Zhaodong Nan, Xia Xue, Wanguo Hou, Xin Yan and Shuhua Han

Page 780

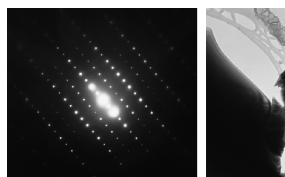


1:1 and 1:2 complexes between CTAB and β -CD have formed as self-assembly templates in the formation of MCM-41.

The structure and ordering of zirconium and hafnium containing garnets studied by electron channelling, neutron diffraction and Mössbauer spectroscopy

Karl R. Whittle, Gregory R. Lumpkin, Frank J. Berry, Gordon Oates, Katherine L. Smith, Sergey Yudintsev and Nestor J. Zaluzec

Page 785



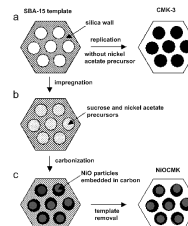
Dynamical diffraction pattern for kimzeyite along the 311 direction, used in the electron channelling analysis, and a bright field image of the sample investigated, the magnification is 25000.

Rapid Communications

Synthesis, characterization, and electrochemical properties of ordered mesoporous carbons containing nickel oxide nanoparticles using sucrose and nickel acetate in a silica template

Yulin Cao, Jieming Cao, Mingbo Zheng, Jinsong Liu and Guangbin Ji

Page 792

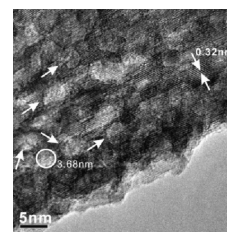


Schematic drawings of synthesis routes for the NiOCMK materials.

Ionic liquid of $[Bmim]^+ Cl^-$ for the preparation of hierarchical nanostructured rutile titania

Ningya Yu, Liming Gong, Huijuan Song, Yong Liu and Donghong Yin

Page 799



Ionic liquid of $[Bmim]^+ Cl^-$ was employed to prepare rutile titania with hierarchical nanostructure, in which the nanorods were interaggregated to fabricate a large mesoporous structure and the voids packed in the nanorods formed a small mesostructure.

Author inquiries

Submissions

For detailed instructions on the preparation of electronic artwork, consult Elsevier's Author Gateway at <http://authors.elsevier.com>.

Other inquiries

Visit Elsevier's Author Gateway (<http://authors.elsevier.com>) for the facility to track accepted articles and set up e-mail alerts to inform you of when an article's status has changed. The Author Gateway also provides detailed artwork guidelines, copyright information, frequently asked questions and more.

Contact details for questions arising after acceptance of an article, especially those relating to proofs, are provided after registration of an article for publication.

Language Polishing

Authors who require information about language editing and copyediting services pre- and post-submission should visit <http://www.elsevier.com/wps/find/authorshome.authors/languagepolishing> or contact authorsupport@elsevier.com for more information. Please note Elsevier neither endorses nor takes responsibility for any products, goods, or services offered by outside vendors through our services or in any advertising. For more information please refer to our Terms & Conditions at http://www.elsevier.com/wps/find/termsconditions.cws_home/termsconditions.

For a full and complete Guide for Authors, please refer to *J. Solid State Chem.*, Vol. 180, Issue 1, pp. *bmi-bmv*. The instructions can also be found at http://www.elsevier.com/wps/find/journaldescription.cws_home/622898/authorinstructions.

Journal of Solid State Chemistry has no page charges.